

Chapter 7 Chemistry

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7. Write a balanced chemical equation for each chemical reaction: a) Solid zinc reacts with aqueous hydrochloric acid (HCl) to form zinc chloride and hydrogen gas. b) Gaseous methane (CH₄) reacts with oxygen gas to form carbon dioxide gas and liquid water.

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number of the isotope

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252Chapter 7 Energy and Chemical Reactions 1 Although chemists recognize that matter can be converted into energy and energy into matter, this matter?energy conversion is small enough to be disregarded. 7.1 Energy253 objeCtive7 objeCtive5 objeCtive7

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Questions. Question 1. Discuss the general characteristics of group 15 elements with reference to their electronic configuration, oxidation state, atomic size, ionisation enthalpy and electro negativity. Answer: General characteristic of group – 15 elements : (A) Electronic configurations :

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